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(CH 3) has an unpaired electron in a non-bonding orbital on the carbon atom, A valence bond description of PF 5 uses resonance between different PF 4 + F

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Valence electron - definition of valence electron

The presence of this magnetic field in substances with unpaired electrons is evident in nitric oxide, boron and oxygen, all of which have one or more unpaired valence

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Natural bond orbitals (NBOs) Each bonding NBO AB (the donor) (the acceptor) to complete the span of the valence space:

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